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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.003 Å R factor = 0.056 wR factor = 0.110 Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. Received 17 April 2006

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10-Benzyl-7,8-dihydro-7,7-dimethyl-5-phenyl-5*H*-indeno[1,2-*b*]quinoline-9,11(6*H*,10*H*)-dione

In the molecule of the title compound, $C_{31}H_{27}NO_2$, the dihydropyridine and cyclohexene rings adopt flattened-boat and envelope conformations, respectively. The molecules are linked by intermolecular $C-H \cdot \cdot \cdot O$ hydrogen bonds, forming a chain structure.

Comment

Multicomponent reactions (MCRs) occupy an outstanding position in organic and medicinal chemistry for their high degree of atom economy, application in combinatorial chemistry and diversity-oriented synthesis (Ramón & Yus, 2005; Burke & Schreiber, 2004; Andreana *et al.*, 2004). It is well known that indenoquinoline derivatives show a diverse range of biological properties, such as antitumour agents, acetyl-cholonesterase inhibitors and potent new cytotoxic agents (Deady *et al.*, 1999, 2000; Rampa *et al.*, 2000; Yamato *et al.*, 1989). In this paper, we report the crystal structure of the title compound, (I).



In the molecule of compound (I) (Fig. 1), the bond lengths and angles are in normal ranges (Allen *et al.*, 1987).

An examination of the deviations from the least-squares planes through the individual rings shows that rings *A* (C7–C12), *B* (C19–C24), *C* (C26–C31) and *D* (C1/C2/C6/C7/C12) are planar. Rings *E* (N1/C1–C5) and *F* (C4/C5/C13–C16) are, of course, not planar. Ring *E* adopts a flattened-boat conformation, with puckering parameters (Cremer & Pople, 1975) $\varphi_2 = 121.56 (2)^\circ$, $\theta_2 = 73.81 (2)^\circ$ and $Q_T = 0.154 (2)$ Å, while the conformation of ring *F* is an envelope [$\varphi_2 = -110.34 (2)^\circ$, $\theta_2 = 24.25 (2)^\circ$ and $Q_T = 0.468 (2)$ Å] with atom C15 at the flap position, 0.625 (3) Å from the mean plane through the other five atoms.

The crystal structure of (I) is stabilized by intermolecular $C-H\cdots O$ hydrogen bonds (Table 1), resulting in the formation of a chain structure (Fig. 2).

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A drawing of the title molecular structure, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms have been omitted for clarity.



Figure 2

A packing diagram of (I). Hydrogen bonds are shown as dashed lines.

Experimental

The title compound was prepared by the reaction of 2-phenylacetaldehyde (120 mg, 1.0 mmol), 3-(phenylamino)-5,5-dimethylcyclohex-2-enone (220 mg, 1.0 mmol) and 1,3-indanedione (150 mg, 1.0 mmol) in a mixed solvent of glycol (0.5 ml) and acetic acid (1.0 ml) under microwave irradiation. Single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a 95% aqueous ethanol solution (yield 380 mg, 85%; m.p. 480-482 K). Spectroscopic analysis: IR (KBr, cm^{-1}): 1686 (CO), 1637 (CO).

Crystal data

C H NO	7 - 4
$C_{31}\Pi_{27}INO_2$	Z = 4
$M_r = 445.54$	$D_x = 1.307 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 14.783 (2) Å	$\mu = 0.08 \text{ mm}^{-1}$
$b = 9.4408 (12) \text{\AA}$	T = 173 (2) K
c = 16.345 (2) Å	Block, red
$\beta = 96.878 \ (3)^{\circ}$	$0.79 \times 0.32 \times 0.14 \text{ mm}$
$V = 2264.6 (5) \text{ Å}^3$	

21579 measured reflections

 $R_{\rm int}=0.040$

 $\theta_{\rm max} = 25.4^{\circ}$

4144 independent reflections

3611 reflections with $I > 2\sigma(I)$

Data collection

Rigaku Mercury diffractometer ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.939, T_{\max} = 0.989$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0325P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.056$	+ 0.9725P]
$wR(F^2) = 0.110$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.18	$(\Delta/\sigma)_{\rm max} < 0.001$
4144 reflections	$\Delta \rho_{\rm max} = 0.20 \text{ e } \text{\AA}^{-3}$
310 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1 Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C14-H14B\cdots O2^{i}$	0.99	2.40	3.337 (2)	157
$C18-H18A\cdots O2^{i}$	0.98	2.58	3.429 (2)	145

Symmetry code: (i) $-x + \frac{3}{2}$, $y - \frac{1}{2}$, $-z + \frac{3}{2}$.

H atoms were positioned geometrically, with C-H = 0.95, 0.98, 0.99 and 1.00 Å for aromatic, methyl, methylene and methine H, respectively, and were constrained to ride on their parent atoms, with $U_{iso}(H) = xU_{eq}(C)$, where x = 1.5 for methyl H and x = 1.2 for all other H.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Andreana, P. R., Liu, C. C. & Schreiber, S. L. (2004). Org. Lett. 6, 4231-4233. Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.

Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

Burke, M. D. & Schreiber, S. L. (2004). Angew. Chem. Int. Ed. 43, 46-58.

Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.

- Deady, L. W., Desneves, J., Kaye, A. J., Finlay, G. J., Baguley, B. C. & Denny, W. A. (2000). *Bioorg. Med. Chem.* 8, 977–984.
- Deady, L. W., Desneves, J., Kaye, A. J., Thompson, M., Finlay, G. J., Baguley, B. C. & Denny, W. A. (1999). *Bioorg. Med. Chem.* 7, 2801– 2809.
- Ramón, D. J. & Yus, M. (2005). Angew. Chem. Int. Ed. 44, 1602–1634.
- Rampa, A., Bisi, A., Belluti, F., Gobbi, S., Valenti, P., Andrisano, V., Cavrini, V., Cavalli, A. & Recanatini, M. (2000). *Bioorg. Med. Chem.* 8, 497–506. Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Yamato, M., Takeuchi, Y., Hashigaki, K., Ikeda, Y., Chang, M. R., Takeuchi, K., Matsushima, M., Tsuruo, T., Tashiro, T., Tsukagoshi, S., Yamashita, Y. & Nakano, H. (1989). J. Med. Chem. 32, 1295–1300.